# **MP1-Structure and Properties of Matter**

Content Area: Science Course(s): Science 5

Time Period: Marking Period 1

Length: MP 1 Status: Published

# **Essential Questions**

- What are the properties of matter and what happens when matter changes state?
- When two substances are mixed, what is formed and is it always something new?

# **Big Ideas**

- Because matter exists as particles that are too small to see, matter is always conserved even if it seems to disappear.
- Measurements of observable properties can be used to identify particular materials.
- Chemical reactions that occur when substances are mixed can be identified by the emergence of substances with different properties; the total mass remains the same.

# **Cross-Curricular Integration**

**Integration Area: Language Arts** 

- RI.5.1 Quote accurately from a text and make relevant connections when explaining what the text says explicitly and when drawing inferences from the text.
- RI.5.7 Draw on information from multiple print or digital sources, demonstrating the ability to locate an answer to a question quickly or to solve a problem efficiently.

### Activity:

Students will research using library resources to find out about atoms. Students will find out who first used the word atom and what an atom is.

# **Science and Society**

**Charles Goodyear** 

**Process Vulcanization** 

#### **Social Justice**

See Social Studies Appendix C

Book: Pigskins to Paintbrushes by Don Tate (National Art Day)

Book: Granddad Budd: A Veteran's Day Story by Sharon Ferry (Veterans Day)

# **CSDT Technology Integration**

8.1.5.DA.1: Collect, organize, and display data in order to highlight relationships or support a claim.

# Activity:

Students will observe, measure, and graph quantities of H2O as a liquid, a solid, and in mixtures to provide evidence that regardless of the type of change that occurs when heating, cooling, or mixing substances, the total weight of matter is conserved.

# **Science and Engineering Practices**

# **Asking Questions and Defining Problems**

• Ask questions about what would happen if a variable is changed.

# **Developing and Using Models**

• Use a model to test cause and effect relationships or interactions concerning the functioning of a natural or designed system.

# **Planning and Carrying Out Investigations**

- Make observations and/or measurements to produce data to serve as the basis for evidence for an explanation of a phenomenon or test a design solution.
- Make predictions about what would happen if a variable changes.
- Test two different models of the same proposed object, tool, or process to determine which better meets criteria for success.

### **Constructing Explanations and Designing Solutions**

• Use evidence (e.g., measurements, observations, patterns) to construct or support an explanation or

design a solution to a problem.

### **Engaging in Argument from Evidence**

• Distinguish among facts, reasoned judgment based on research findings, and speculation in an explanation.

## Construct and/or support an argument with evidence, data, and/or a model

• Use data to evaluate claims about cause and effect.

# Obtaining, Evaluating, and Communicating Information

- Read and comprehend grade-appropriate complex texts and/or other reliable media to summarize and obtain scientific and technical ideas and describe how they are supported by evidence.
- Compare and/or combine across complex texts and/or other reliable media to support the engagement in other scientific and/or engineering practices.
- Communicate scientific and/or technical information orally and/or in written formats, including various forms of media and may include tables, diagrams, and charts.

# **Enduring Understandings**

### **Matter and its Interactions**

- 5-PS1-1 Develop a model to describe that matter is made of particles too small to be seen.
- 5-PS1-2 Measure and graph quantities to provide evidence that regardless of the type of change that occurs when heating, cooling, or mixing substances, the total weight of matter is conserved.
- 5-PS1-3 Make observations and measurements to identify materials based on their properties.
- 5-PS1-4 Conduct an investigation to determine whether the mixing of two or more substances results in new substances.

### **Student Learning Standards**

### **ELA/Literacy**

- RI.5.7 Draw on information from multiple print or digital sources, demonstrating the ability to locate an answer to a question quickly or to solve a problem efficiently. (5-PS1-1)
- W.5.7 Conduct short research projects that use several sources to build knowledge through investigation of different aspects of a topic. (5-PS1-2),(5-PS1-3),(5-PS1-4)
- W.5.8 Recall relevant information from experiences or gather relevant information from print and digital sources; summarize or paraphrase information in notes and finished work, and provide a list of sources. (5-PS1-2),(5-PS1-3),(5-PS1-4)

W.5.9 Draw evidence from literary or informational texts to support analysis, reflection, and research. (5-PS1-2),(5-PS1-3),(5-PS1-4)

#### **Mathematics**

- MP.2 Reason abstractly and quantitatively. (5-PS1-1),(5-PS1-2),(5-PS1-3) MP.4 Model with mathematics. (5-PS1-1),(5-PS1-2),(5-PS1-3)
- MP.5 Use appropriate tools strategically. (5-PS1-2),(5-PS1-3)
- 5.NBT.A.1 Explain patterns in the number of zeros of the product when multiplying a number by powers of 10, and explain patterns in the placement of the decimal point when a decimal is multiplied or divided by a power of 10. Use whole-number exponents to denote powers of 10. (5-PS1-1)
- 5.NF.B.7 Apply and extend previous understandings of division to divide unit fractions by whole numbers and whole numbers by unit fractions. (5-PS1-1)
- 5.MD.A.1 Convert among different-sized standard measurement units within a given measurement system (e.g., convert 5 cm to 0.05 m), and use these conversions in solving multi-step, real-world problems. (5-PS1-2)
- 5.MD.C.3 Recognize volume as an attribute of solid figures and understand concepts of volume measurement. (5-PS1-1)
- 5.MD.C.4 Measure volumes by counting unit cubes, using cubic cm, cubic in, cubic ft., and non-standard units. (5-PS1-1)

#### **Focus Areas**

#### Knowledge

- Matter is a term that applies to all of the stuff around us and it is made of particles that are too small to see.
- When substances are heated, cooled, or mixed the total weight before and after is always the same.
- Substances can be identified based on observable and measureable properties.
- Sometimes when two substances are mixed, each of the substances keeps its original properties and sometimes a new substance is formed.

#### **Skills**

- Give an example of what is matter.
- Describe how gases are made from matter particles that are too small to be seen. (Ex: an inflated balloon)
- Measure and graph the weights of matter before and after being heated, cooled, or mixed.
- Identify materials based on various observable properties.
- Determine whether the mixing of two substances always results in the formation of new substances or not and provide examples.
- Identify the differences between soluble and insoluble solutions.

### **Understandings**

- Develop a model to describe that matter is made of particles too small to be seen.
- Measure and graph quantities to provide evidence that regardless of the type of change that occurs when heating, cooling.
- Make observations and measurements to identify materials based on their properties.
- Conduct an investigation to determine whether the mixing of two or more substances results in new substances.

#### **Resources**

# **Primary Resource**

Scott Foresman Interactive Science, Pearson, 2016

• Properties of Matter - Chapter 1

#### Leveled Readers

- *Matter and Its Properties*
- Properties of Matter
- Pioneers of Physics
- Changes in Matter
- Baking Chemistry

# **Scientific Inquiry**

#### Core

- Properties of solids, liquids and gasses inquiry lab
- Conservation of matter and chemical reaction inquiry lab
- Particles in matter lab

### Supplemental

- How are weight and volume affected when objects are combined? p.2
- What are some ways to separate a mixture? p. 40