

Unit 12: Thermochemistry Copied from: Chemistry, Copied on: 06/05/25

Content Area: **Template**

Course(s):

Time Period:

Length:

Status: **Published**

State Mandated Topics Addressed in this Unit

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N/A	N/A

Unit 12: Thermochemistry

Learning Objectives

- How can you determine the amount of energy absorbed or released in a chemical or physical process?
- How is energy absorbed or released in a reaction?
- How is energy conserved in a chemical or physical process?
- What is meant by “energy is conserved”?

Essential Skills

- Define heat capacity, enthalpy, entropy, thermochemical equation, Hess’s law.
- Define system and surroundings; give examples.
- Describe how you can measure the change of enthalpy in a reaction
- Explain the difference between energy and heat
- Express the enthalpy change for a given reaction
- Identify 2 factors on which the heat capacity of an object depends
- Write and solve thermochemical equations for enthalpy

Standards

9-12.HS-PS1-3

Plan and conduct an investigation to gather evidence to compare the structure of substances at the bulk scale to infer the strength of electrical forces between particles.

9-12.HS-PS3-4

Plan and conduct an investigation to provide evidence that the transfer of thermal energy

when two components of different temperature are combined within a closed system results in a more uniform energy distribution among the components in the system (second law of thermodynamics).

9-12.HS-PS3-1

Create a computational model to calculate the change in the energy of one component in a system when the change in energy of the other component(s) and energy flows in and out of the system are known.

9-12.HS-PS1-4

Develop a model to illustrate that the release or absorption of energy from a chemical reaction system depends upon the changes in total bond energy.

Instructional Tasks/Activities

- Common assessment chapter test
- Common assessment quiz
- Constructed response
- Demolition Derby Activity and Kinetics
- Do now's and/or exit slips
- Energy flow and systems
- Entropy and Thermodynamic Systems Room Activity
- Graphic organizers or models
- Guided practice
- Heat, Work, and Thermodynamics Demonstrations
- Homework
- Individual, small, and large group work
- Kinetics lab
- Laboratory investigations within small groups
- Review Activity
- Thermodynamic Systems Observations Outdoors
- Thermodynamics Intro and Questions
- Thermodynamics systems and Earth's Climate
- Weather, Climate and Energy

Assessment Procedure

- Classroom Total Participation Technique
- Classwork
- DBQ
- Essay
- Exit Ticket/Entrance Ticket/Do Now
- Flashcards and/or drill and practice
- Inquiry based activities with reflective discussion
- Journal / Student Reflection

- Kahoot
- Laboratory groups
- Lecture with note taking or guided notes
- Online models and simulators
- Other named in lesson
- Peer Review
- Performance
- Power Point Presentation
- Problem Correction
- Project
- Quiz
- Rubric
- Teacher Collected Data
- Test
- Whole and small group discussions
- Worksheet

Recommended Technology Activities

- Appropriate Content Specific Online Resource
- Chromebook
- Copy/Paste Content Specific Link Here
- Copy/Paste Content Specific Link Here
- Copy/Paste Content Specific Link Here
- Gimkit
- GoGuardian
- Google Classroom
- Google Docs
- Google Forms
- Google Slides
- Kahoot
- MagicSchool AI
- Other- Specified in Lesson
- Quiziz
- Screencastify

Accommodations & Modifications & Differentiation

Accommodations and Modifications should be used to meet individual needs. Their IEP and 504 plans should be used in addition to the following suggestions.

Gifted and Talented

- Compare & Contrast
- Conferencing
- Debates
- Jigsaw
- Peer Partner Learning
- Problem Solving
- Structured Controversy
- Think, Pair, Share
- Tutorial Groups

Instruction/Materials

- alter format of materials (type/highlight, etc.)
- color code materials
- eliminate answers
- extended time
- extended time
- large print
- modified quiz
- modified test
- Modify Assignments as Needed
- Modify/Repeat/Model directions
- necessary assignments only
- Other (specify in plans)
- other- named in lesson
- provide assistance and cues for transitions
- provide daily assignment list
- read class materials orally
- reduce work load
- shorten assignments
- study guide/outline
- utilize multi-sensory modes to reinforce instruction

Environment

- alter physical room environment
- assign peer tutors/work buddies/note takers
- assign preferential seating
- individualized instruction/small group
- modify student schedule (Describe)
- other- please specify in plans
- provide desktop list/formula

Honors Modifications

Resources

- Resource 1
- Resource 2
- Resource 3
- Resource 4
- Resource 5