# **Unit 4: Stoichiometry & Types of Reactions**

Content Area: **Template** 

Course(s): Time Period: Length:

Status: Published

**State Mandated Topics Addressed in this Unit** 

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N/A	N/A

# **Unit 4: Stoichiometry & Types of Reactions**

#### **Essential Questions**

- How do net ionic equations reveal the active species in solution?
- How do you translate between moles, mass, and particles?
- How does reaction type inform product prediction?
- What is the significance of the limiting reactant?
- Why does actual yield differ from theoretical yield?
- · Why must chemical equations be balanced?

#### **Standards**

SCI.HS-PS1-7	Use mathematical representations to support the claim that atoms, and therefore mass, are conserved during a chemical reaction.
9-12.HS-PS1-3	Plan and conduct an investigation to gather evidence to compare the structure of substances at the bulk scale to infer the strength of electrical forces between particles.
9-12.HS-PS1-1	Use the periodic table as a model to predict the relative properties of elements based on the patterns of electrons in the outermost energy level of atoms.
9-12.HS-PS1-2	Construct and revise an explanation for the outcome of a simple chemical reaction based on the outermost electron states of atoms, trends in the periodic table, and knowledge of the patterns of chemical properties.

#### **Assessment Procedure**

- Classroom Total Participation Technique
- Classwork

- DBQ
- Essay
- Exit Ticket/Entrance Ticket/Do Now
- Flashcards and/or drill and practice
- Inquiry based activities with reflective discussion
- Journal / Student Reflection
- Kahoot
- Laboratory groups
- Lecture with note taking or guided notes
- Online models and simulators
- Other named in lesson
- Peer Review
- Performance
- Power point presentation
- Problem Correction
- Project
- Quiz
- Rubric
- Teacher Collected Data
- Test
- Whole and small group discussions
- Worksheet

# **Recommended Technology Activities**

- Appropriate Content Specific Online Resource
- Chromebook
- Copy/Paste Content Specific Link Here
- Copy/Paste Content Specific Link Here
- Copy/Paste Content Specific Link Here
- Gimkit
- GoGuardian
- Google Classroom
- Google Docs
- Google Forms
- Google Slides
- Kahoot
- MagicSchool Al
- Other- Specified in Lesson

- Quiziz
- Screencastify

#### **Accommodations & Modifications & Differentiation**

Accommodations and Modifications should be used to meet individual needs. Their IEP and 504 plans should be used in addition to the following suggestions.

#### **Gifted and Talented**

- Compare & Contrast
- Conferencing
- Debates
- Jigsaw
- Peer Partner Learning
- · Problem Solving
- Structured Controversy
- Think, Pair, Share
- Tutorial Groups

## **Instruction/Materials**

- alter format of materials (type/highlight, etc.)
- color code materials
- · eliminate answers
- extended time
- extended time
- large print
- · modified quiz
- modified test
- · Modify Assignments as Needed
- Modify/Repeat/Model directions
- necessary assignments only
- Other (specify in plans)
- other- named in lesson
- provide assistance and cues for transitions
- provide daily assignment list
- · read class materials orally

- reduce work load
- shorten assignments
- study guide/outline
- utilize multi-sensory modes to reinforce instruction

## **Environment**

- alter physical room environment
- assign peer tutors/work buddies/note takers
- assign preferential seating
- individualized instruction/small group
- modify student schedule (Describe)
- other- please specify in plans
- provide desktop list/formula

#### **Honors Modifications**

#### Resources

- Resource 1
- Resource 2
- Resource 3
- Resource 4
- Resource 5