

# Unit 3: Processing Matter: Naming, Reactions and their Rates, Equilibrium Effects

Content Area: **Science**  
Course(s): **Chemistry/Lab Honors**  
Time Period: **2nd Marking Period**  
Length: **8.5 Weeks**  
Status: **Published**

## Unit Overview

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Many substances react chemically with other substances to form new substances with different properties. This change in properties results from the ways in which atoms from the original substances are combined and rearranged in the new substances.

## Transfer

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Students will be able to independently use their learning to interpret and predict how substances react with each other.

## Meaning

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## Understandings

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- 1) There are different classes of molecules, each with their own system of naming.
- 2) Chemical equations are used to represent chemical reactions and show that mass can neither be created nor destroyed.
- 3) There are different types of chemical reactions that we observe in everyday life.
- 4) The rate of a reaction is influenced by several factors.
- 5) All reactions work toward equilibrium.

## Essential Questions

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- 1) How does chemical naming exhibit organizational patterns?
- 2) Why must the mass of reactants equal the mass of products in a chemical reaction?
- 3) What characteristics are used to classify chemical reactions?
- 4) How does collision theory explain the factors affecting reaction rate?
- 5) How does nature correct unbalance?
- 6) How do chemical reactions attain and maintain a state of equilibrium

## **Application of Knowledge and Skill**

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### **Students will know...**

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- 1) How to name Ionic, Covalent Compounds and Acids.
- 2) How to accurately represent and predict the type of chemical reactions.
- 3) How to balance chemical equations using coefficients.
- 4) Predict how a reaction rate will be influenced.
- 5) Predict how an equilibrium will be established and how the system will react when changed.

### **Students will be skilled at...**

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- 1) Comparing and contrasting different chemical compounds and naming systems
- 2) Analyze a reaction to see if the predicted results are obtained
- 3) Prove that a balanced reaction satisfies the law of conservation of mass
- 4) Construct a balanced chemical reaction given reactants.
- 5) Develop a logical argument as to how different factors affect the rate of a reaction based upon collision theory.
- 6) Analyzing Q and K to determine the direction a reaction must shift to reach equilibrium.
- 7) Utilizing LeChatlier's Principle to predict behaviors in chemical reactions.

## Vocabulary

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Academic Vocabulary	Application Vocabulary
acid	analyze
activity series	balance (verb)
chemical reaction	coefficient
combustion	contrast
composition	prediction
covalent	rearrangement
decomposition	catalyst
double replacement	collision theory
ionic	enzyme
monatomic	frequency
polyatomic	barrier
product	absorption
reactant	LeChatlier's Principle
single replacement	
concentration	
endothermic	
exothermic	
energy	
equilibrium	
equilibrium constant	
reaction quotient	
reaction rate	
surface area	

## Learning Goal 1

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Students will demonstrate knowledge of the correct name and formula for different classes of compounds.

### Proficiency Scale

- Students will demonstrate knowledge of the correct name and formula for different classes of compounds.

SCI.HS-PS1-2

Construct and revise an explanation for the outcome of a simple chemical reaction based on the outermost electron states of atoms, trends in the periodic table, and knowledge of the patterns of chemical properties.

## Target 1

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SWBAT connect name and formula for given monatomic and polyatomic ions.

- SWBAT connect name and formula for given monatomic and polyatomic ions.

## **Target 2**

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SWBAT connect correct formulas of both ionic and covalent compounds.

- SWBAT connect correct formulas of both ionic and covalent compounds.

## **Target 3**

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SWBAT connect names and correct formula of an acid.

- SWBAT connect names and correct formula of an acid.

## **Learning Goal 2**

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Students will be able to express chemical changes qualitatively by completing and balancing chemical reactions.

## **Proficiency Scale**

- Students will be able to express chemical changes qualitatively by completing and balancing chemical reactions.

SCI.HS-PS1-2

Construct and revise an explanation for the outcome of a simple chemical reaction based on the outermost electron states of atoms, trends in the periodic table, and knowledge of the patterns of chemical properties.

## **Target 1**

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SWBAT formulate skeleton equations using coefficients.

- SWBAT formulate skeleton equations using coefficients.

## **Target 2**

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SWBAT accurately create a chemical reactions using word equations and formula equations.

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### **Target 3**

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SWBAT classify chemical reactions as one of the 5 basic types of reactions.

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### **Target 4**

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SWBAT prove whether a given reaction will occur or not, using the activity series.

- SWBAT prove whether a given reaction will occur or not, using the activity series.

### **Target 5**

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SWBAT synthesize the product(s) and type of reaction after looking at the reactant(s) in a reaction; writing the complete balanced reaction.

- SWBAT synthesize the product(s) and type of reaction after looking at the reactant(s) in a reaction; writing the complete balanced reaction.

### **Target 6**

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SWBAT write net ionic reactions given molecular formulas of reactants.

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### **Learning Goal 3**

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Connect the frequency and success of molecular collisions to the rate of a reaction, considering that there needs to be sufficient energy inputed to overcome the activation energy barrier.

### **Proficiency Scale**

- Connect the frequency and success of molecular collisions to the rate of a reaction, considering that there needs to be sufficient energy inputed to overcome the activation energy barrier.

SCI.HS-PS3-1

Create a computational model to calculate the change in the energy of one component in a system when the change in energy of the other component(s) and energy flows in and out of the system are known.

SCI.HS-PS1-4

Develop a model to illustrate that the release or absorption of energy from a chemical reaction system depends upon the changes in total bond energy.

SCI.HS-PS1-5

Apply scientific principles and evidence to provide an explanation about the effects of changing the temperature or concentration of the reacting particles on the rate at which a reaction occurs.

SCI.HS-PS3-2

Develop and use models to illustrate that energy at the macroscopic scale can be accounted for as a combination of energy associated with the motions of particles (objects) and energy associated with the relative position of particles (objects).

### **Target 1**

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SWBAT differentiate between the four factors that influence reaction rate and hypothesize changes in rates when one of those factors are changed.

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### **Target 2**

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SWBAT design energy diagrams and critique them based on one of the four factors affecting rate being changed.

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### **Learning Goal 4**

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Students will be able to establish an equilibrium and predict changes in equilibrium based on LeChatelier's principle.

### **Proficiency Scale**

- Students will be able to establish an equilibrium and predict changes in equilibrium based on LeChatelier's principle.

SCI.HS-PS1-6

Refine the design of a chemical system by specifying a change in conditions that would produce increased amounts of products at equilibrium.

### **Target 1**

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SWBAT formulate the Equilibrium Constant Expression.

- SWBAT formulate the Equilibrium Constant Expression.

### **Target 2**

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SWBAT formulate the Equilibrium Constant given the amounts of reactants and products at equilibrium.

- SWBAT formulate the Equilibrium Constant given the amounts of reactants and products at equilibrium.

### **Target 3**

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SWBAT use concepts to solve non-routine problems that occur when disturbing equilibrium, predicting changes and shifts.

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### **Formative Assessment and Performance Opportunities**

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Lab Reports

Worksheets

PowerPoints with Notes

Homework and Classwork Activities

Group Activities

In Class Discussion

- Lab Reports, Worksheets, PowerPoints with Notes, Homework and Classwork Activities, Group Activities, In Class Discussion

### **Summative Assessment**

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Unit Assessment will be created collaboratively and used for every student in the course. In addition, there will be other assessments in the form of lab reports, pen and paper tests, and quizzes. Common Assessment is administered through LinkIt.

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### **Accommodations/Modifications**

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All instruction, labs, activities, and assessments will be modified and enhanced to adhere to individual student's IEPs and 504s. As well differentiated classroom management strategies will be utilized as to adhere to these students individual plans as well. These strategies can be but are not limited to the following: Graphic Organizers and Flow Charts to help visualization of content, Using acronyms and mnemonic devices to assist memorization, Academic Games to practice skills and deepen understanding of topics, Stations and Learning Centers to influence peer tutoring, Utilizing Reviews and Study Guides to reteach complex content before summative and formative assessment, Providing additional resources to students during and outside of school (videos, study guides, teacher notes, web-based resources).

**On Assessments, College Prep Chemistry Students will have access to the following resources: Naming Flow Chart, Reaction Examples and Names of Reaction Types.**

**On Assessments, Honors AND College Prep Chemistry Students will have access to the following resources: Periodic Table**

- All instruction, labs, activities, and assessments will be modified and enhanced to adhere to individual student's IEPs and 504s. As well differentiated classroom management strategies will be utilized as to adhere to these students individual plans as well. These strategies can be but are not limited to the following: Graphic Organizers and Flow Charts to help visualization of content, Using acronyms and mnemonic devices to assist memorization, Academic Games to practice skills and deepen understanding of topics, Stations and Learning Centers to influence peer tutoring, Utilizing Reviews and Study Guides to reteach complex content before summative and formative assessment, Providing additional resources to students during and outside of school (videos, study guides, teacher notes, web-based resources). On Assessments, College Prep Chemistry Students will have access to: Naming Flow Chart, Reaction Examples and Names of Reaction Types.

## **Unit Resources**

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Teacher generated PowerPoints, Notes, Labs and Worksheets

Textbooks

Resource Books

Internet Resources

Computer Based Activities

Lab for this unit: Conductivity Lab, Naming and formula writing lab, Activity series lab, Precipitate lab, Penny -Single replacement lab- Activity series lab, Reaction Rates, Equilibrium lab

- Teacher generated PowerPoints, Notes, Labs and Worksheets, Textbooks, Resource Books, Internet Resources, Computer Based Activities

## **21st Century Life and Careers**

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CRP.K-12.CRP1.1

Career-ready individuals understand the obligations and responsibilities of being a member of a community, and they demonstrate this understanding every day through their interactions with others. They are conscientious of the impacts of their decisions on others and the environment around them. They think about the near-term and long-term consequences of their actions and seek to act in ways that contribute to the betterment of their teams, families, community and workplace. They are reliable and consistent in going beyond the minimum expectation and in participating in activities that serve the greater good.

CRP.K-12.CRP2.1

Career-ready individuals readily access and use the knowledge and skills acquired through experience and education to be more productive. They make connections between abstract concepts with real-world applications, and they make correct insights about when it is appropriate to apply the use of an academic skill in a workplace situation.



CRP.K-12.CRP4.1	Career-ready individuals communicate thoughts, ideas, and action plans with clarity, whether using written, verbal, and/or visual methods. They communicate in the workplace with clarity and purpose to make maximum use of their own and others' time. They are excellent writers; they master conventions, word choice, and organization, and use effective tone and presentation skills to articulate ideas. They are skilled at interacting with others; they are active listeners and speak clearly and with purpose. Career-ready individuals think about the audience for their communication and prepare accordingly to ensure the desired outcome.
CRP.K-12.CRP5.1	Career-ready individuals understand the interrelated nature of their actions and regularly make decisions that positively impact and/or mitigate negative impact on other people, organization, and the environment. They are aware of and utilize new technologies, understandings, procedures, materials, and regulations affecting the nature of their work as it relates to the impact on the social condition, the environment and the profitability of the organization.
CRP.K-12.CRP6.1	Career-ready individuals regularly think of ideas that solve problems in new and different ways, and they contribute those ideas in a useful and productive manner to improve their organization. They can consider unconventional ideas and suggestions as solutions to issues, tasks or problems, and they discern which ideas and suggestions will add greatest value. They seek new methods, practices, and ideas from a variety of sources and seek to apply those ideas to their own workplace. They take action on their ideas and understand how to bring innovation to an organization.
CRP.K-12.CRP7.1	Career-ready individuals are discerning in accepting and using new information to make decisions, change practices or inform strategies. They use reliable research process to search for new information. They evaluate the validity of sources when considering the use and adoption of external information or practices in their workplace situation.
CRP.K-12.CRP8.1	Career-ready individuals readily recognize problems in the workplace, understand the nature of the problem, and devise effective plans to solve the problem. They are aware of problems when they occur and take action quickly to address the problem; they thoughtfully investigate the root cause of the problem prior to introducing solutions. They carefully consider the options to solve the problem. Once a solution is agreed upon, they follow through to ensure the problem is solved, whether through their own actions or the actions of others.
CRP.K-12.CRP9.1	Career-ready individuals consistently act in ways that align personal and community-held ideals and principles while employing strategies to positively influence others in the workplace. They have a clear understanding of integrity and act on this understanding in every decision. They use a variety of means to positively impact the directions and actions of a team or organization, and they apply insights into human behavior to change others' action, attitudes and/or beliefs. They recognize the near-term and long-term effects that management's actions and attitudes can have on productivity, morals and organizational culture.
CRP.K-12.CRP11.1	Career-ready individuals take personal ownership of their own education and career goals, and they regularly act on a plan to attain these goals. They understand their own career interests, preferences, goals, and requirements. They have perspective regarding the pathways available to them and the time, effort, experience and other requirements to pursue each, including a path of entrepreneurship. They recognize the value of each step in the education and experiential process, and they recognize that nearly all career paths require ongoing education and experience. They seek counselors, mentors, and other experts to assist in the planning and execution of career and personal goals.
CRP.K-12.CRP12.1	Career-ready individuals positively contribute to every team, whether formal or informal. They apply an awareness of cultural difference to avoid barriers to productive and positive interaction. They find ways to increase the engagement and contribution of all team members. They plan and facilitate effective team meetings.

## Interdisciplinary Connections

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LA.SL.11-12.5	Make strategic use of digital media (e.g., textual, graphical, audio, visual, and interactive elements) in presentations to enhance understanding of findings, reasoning, and evidence and to add interest.
LA.RST.9-10.2	Determine the central ideas, themes, or conclusions of a text; trace the text's explanation or depiction of a complex process, phenomenon, or concept; provide an accurate summary of the text.
LA.RST.9-10.3	Follow precisely a complex multistep procedure when carrying out experiments, taking measurements, or performing technical tasks, attending to special cases or exceptions defined in the text.
LA.RST.9-10.4	Determine the meaning of symbols, key terms, and other domain-specific words and phrases as they are used in a specific scientific or technical context relevant to grades 9-10 texts and topics.
LA.RST.9-10.5	Analyze the relationships among concepts in a text, including relationships among key terms (e.g., force, friction, reaction force, energy).
LA.RST.9-10.7	Translate quantitative or technical information expressed in words in a text into visual form (e.g., a table or chart) and translate information expressed visually or mathematically (e.g., in an equation) into words.
LA.RST.9-10.9	Compare and contrast findings presented in a text to those from other sources (including their own experiments), noting when the findings support or contradict previous explanations or accounts.
LA.RST.11-12.1	Accurately cite strong and thorough evidence from the text to support analysis of science and technical texts, attending to precise details for explanations or descriptions.
LA.RST.11-12.2	Determine the central ideas, themes, or conclusions of a text; summarize complex concepts, processes, or information presented in a text by paraphrasing them in simpler but still accurate terms.
LA.RST.11-12.7	Integrate and evaluate multiple sources of information presented in diverse formats and media (e.g., quantitative data, video, multimedia) in order to address a question or solve a problem.
LA.WHST.9-10.2	Write informative/explanatory texts, including the narration of historical events, scientific procedures/experiments, or technical processes.
LA.WHST.9-10.2.D	Use precise language and domain-specific vocabulary to manage the complexity of the topic and convey a style appropriate to the discipline and context as well as to the expertise of likely readers.
LA.WHST.9-10.7	Conduct short as well as more sustained research projects to answer a question (including a self-generated question) or solve a problem; narrow or broaden the inquiry when appropriate; synthesize multiple sources on the subject, demonstrating understanding of the subject under investigation.
LA.WHST.9-10.9	Draw evidence from informational texts to support analysis, reflection, and research.
LA.WHST.11-12.5	Develop and strengthen writing as needed by planning, revising, editing, rewriting, or trying a new approach, focusing on addressing what is most significant for a specific purpose and audience.
LA.WHST.11-12.7	Conduct short as well as more sustained research projects to answer a question (including a self-generated question) or solve a problem; narrow or broaden the inquiry when appropriate; synthesize multiple sources on the subject, demonstrating understanding of the subject under investigation.
LA.WHST.11-12.8	Gather relevant information from multiple authoritative print and digital sources, using advanced searches effectively; assess the strengths and limitations of each source in terms of the specific task, purpose, and audience; integrate information into the text selectively

	to maintain the flow of ideas, avoiding plagiarism and overreliance on any one source and following a standard format for citation.
LA.WHST.11-12.9	Draw evidence from informational texts to support analysis, reflection, and research.
MA.N-Q.A	Reason quantitatively and use units to solve problems.
MA.N-Q.A.1	Use units as a way to understand problems and to guide the solution of multi-step problems; choose and interpret units consistently in formulas; choose and interpret the scale and the origin in graphs and data displays.
MA.N-Q.A.2	Define appropriate quantities for the purpose of descriptive modeling.
MA.N-Q.A.3	Choose a level of accuracy appropriate to limitations on measurement when reporting quantities.
MA.K-12.1	Make sense of problems and persevere in solving them.
MA.K-12.2	Reason abstractly and quantitatively.
MA.K-12.4	Model with mathematics.
MA.K-12.5	Use appropriate tools strategically.
MA.K-12.6	Attend to precision.
MA.A-CED.A.1	Create equations and inequalities in one variable and use them to solve problems.
MA.A-CED.A.2	Create equations in two or more variables to represent relationships between quantities; graph equations on coordinate axes with labels and scales.
MA.A-CED.A.4	Rearrange formulas to highlight a quantity of interest, using the same reasoning as in solving equations.
MA.A-REI.A.1	Explain each step in solving a simple equation as following from the equality of numbers asserted at the previous step, starting from the assumption that the original equation has a solution. Construct a viable argument to justify a solution method.
MA.A-REI.B.3	Solve linear equations and inequalities in one variable, including equations with coefficients represented by letters.
MA.A-SSE.A.2	Use the structure of an expression to identify ways to rewrite it. For example, see $x^4 - y^4$ as $(x^2)^2 - (y^2)^2$ , thus recognizing it as a difference of squares that can be factored as $(x^2 - y^2)(x^2 + y^2)$ .
MA.A-SSE.A.1b	Interpret complicated expressions by viewing one or more of their parts as a single entity.